12006 Olivine Basalt 206.4 grams

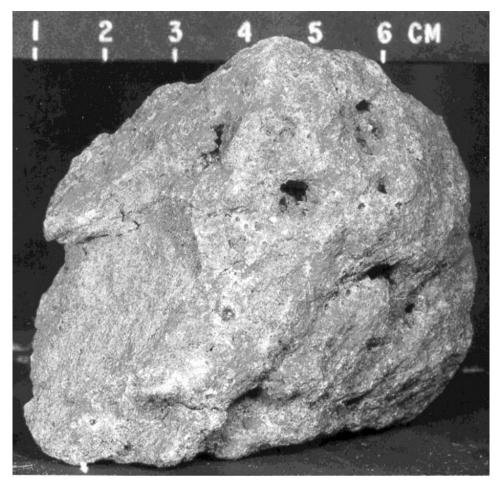


Figure 1: Photo of 12006. Note cavities (vugs) and zap pits. NASA #S69-62339. Scale is shown.

Introduction

James and Wright (1972) originally classified 12006 as a "subophitic feldspathic basalt", while Rhodes et al. (1977) and Neal et al. (1994) termed it an "olivine basalt" based on its chemical composition.

According to Hörz and Hartung (1971), rock 12006 displayed various micrometeorite crater densities on all surfaces, indicating multiple orientations during its history on the lunar surface. This basalt also has several cavities or "vugs" (figure 1).

Mineralogical Mode for 12006

	Neal et
	al. 1994
Olivine	22.6
Pyroxene	28.7
Plagioclase	43.4
Ilmenite	1.1
Chromite +Usp	3.4
Mesostasis	0.5

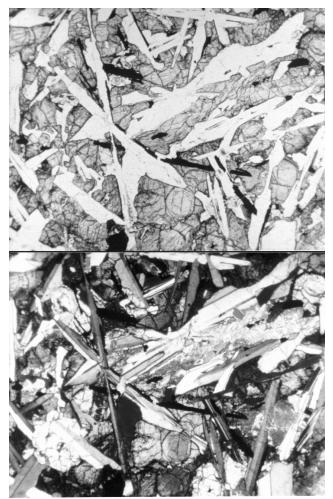


Figure 2: Texture of lunar basalt 12006. Photomicrographs S70-16788-789. Field of view is 2.2 mm. Lower photo is with crossed-nicols.

1000 Neal et al. 1994 Rhodes et al. 1977 La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Tm Yb Lu

Figure 3: Normalized rare-earth-element diagram for 12006.

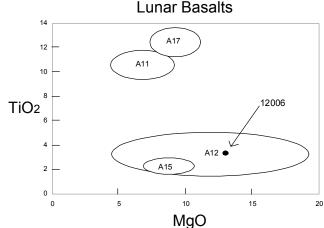


Figure 4: Composition of 12006 compared with other lunar basalts.

Petrography

For some reason, 12006 has not been well studied (although there are several thin sections). The texture is that of a medium-grained subophitic basalt (figure 2). According to Neal et al. (1994), this basalt has a high proportion of plagioclase (43%).

Chemistry

The REE content determined by Neal et al. (1994) did not agree with that of Rhodes et al. (1977) (table 1, figure 3). Neal et al. also found relatively high Na₂O.

Radiogenic age dating

This sample has not been dated (although it was allocated to Wasserburg).

The main mass, 12006,1 is on public display in Japan (figure 5).

List of Photo #s

\$70-16788-789 TS \$70-49155-158 TS \$70-49957-958 TS \$76-27137-143 color mug \$86-38616-618 \$90-33265-266

Table 1. Chemical composition of 12006.

reference weight	Neal 94		Rhodes77		
SiO2 % TiO2 % Al2O3 FeO MnO MgO CaO Na2O K2O P2O5 S % sum	.238 g 3.2 10.6 20.8 0.258 12.8 9.8 0.409 0.075	(a) (a) (a) (a) (a) (a) (a)	44.23 2.59 7.67 20.94 0.29 14.67 8.13 0.2 0.05 0.05	(c) (c) (c) (c) (c) (c) (a) (c) (c)	
Sc ppm V Cr Co Ni Cu Zn Ga Ge ppb As Se	47.2 172 3890 45.2 70	(a) (a) (a) (a)	40.1 6250 60 110	(a) (a) (a) (a)	
Rb Sr Y Zr Nb Mo Ru Rh Pd ppb Ag ppb Cd ppb In ppb Sn ppb Sb ppb Te ppb	104	(a)	89 31 97 6.4	(c) (c) (c)	
Cs ppm Ba La Ce	117 9 26.3	(a) (a) (a)	56 15.7	(b) (a)	
Pr Nd Sm Eu Gd	19.6 5.7 1.42	(a) (a) (a)	3.77 0.72	(a) (a)	
Tb Dy Ho Er	1.35 8.1	(a) (a)	1.02	(a)	
Tm Yb Lu Hf Ta W ppb Re ppb Os ppb Ir ppb Pt ppb	4.1 0.59 3.8 0.56	(a) (a) (a) (a)	3.3 0.47 3	(a) (a) (a)	
Au ppb Th ppm U ppm	0.74	(a)			
technique (a) INAA, (b) IDMS, (c) XRF					

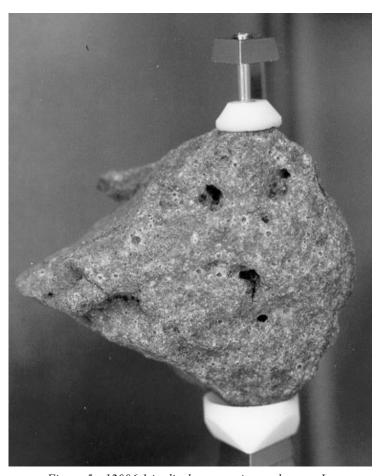


Figure 5: 12006,1 in display mounting on loan to Japan. NASA photo # S90-33266.

